

## HW Solutions # 4

2.8) a) 21, corresponding to the #  
of ways of splitting up  
20 units of energy between  
two systems

$$\begin{matrix} 0, 20 & ; & 1, 19 & ; & \dots & ; & 20, 0 \\ A & B & & A & B & & A & B \end{matrix}$$

1      2       $\dots$       21

b)  $\Delta_b(20, 20) = \binom{20+20-1}{20} = \binom{39}{20}$

$$(2A) \approx \binom{e \cdot 20}{20} = e^{20} \approx 10^{20} / 2 \cdot 3$$

$\frac{1}{2} \times 10^{20} \approx 69 \text{ Billion}$

This is the ways of dividing our  
20 energy units among the combined  
20 oscillators.

$$\begin{aligned} c) N_{\text{total}} &= N_A N_B = \binom{20+10-1}{20} \times \binom{10-1}{0} \\ &= \binom{20+10-1}{20} \binom{10-1}{0} \\ &= \left(\frac{29}{20}\right) \left(\frac{9}{0}\right) = \binom{29}{20} \cdot 1 \end{aligned}$$

$\approx 10 \text{ million}$

d)  $\left(\frac{\frac{N_A}{10} \frac{g_A}{10}}{10+10-1}\right) \left(\frac{\frac{N_B}{10} \frac{g_B}{10}}{10+10-1}\right) = \left(\frac{19}{10}\right)^2 \approx 8.5 \text{ billion}$

e) if we minimize the combined system w/ a loss in entropy in energy transfer between systems A + B, then the system will irreversibly transfer energy to even out it out, i.e. minimizes ~~pro~~entropy.

2) According to the "statistical 2<sup>nd</sup> law of T.D.", a system will be in equilibrium when its macrostate is the one w/ the largest number of microstates.

For a system w/ two subsystems A + B in thermal contact i.e. can exchange energy], the total number of microstates is the product of their individual microstates:

$$N_{\text{tot}} = N_A \cdot N_B$$

If we transfer a small amount of energy  $du$  ~~from~~<sup>to</sup> system A to from system B, then the change in the total number of states is:

$$\begin{aligned} & N_A(u_A + du) N_B(u_B - du) \\ &= [N_A(u_A) + \frac{dN_A}{du} (u_A) du] \\ &\quad \cdot [N_B(u_B) + \frac{dN_B}{du} (u_B) (-du)] \end{aligned}$$

in limit as  $du \rightarrow 0$ :

$$= N_A(u_A) N_B(u_B) + [N'_A N_B - N_A N'_B] du$$

so the change in the total # of microstates - is :

$$[n_A \ln \lambda_B - n_B \ln \lambda_A] du$$

$$= \left[ \frac{n_A}{\bar{n}_A} \ln \lambda_B - \frac{n_B}{\bar{n}_B} \ln \lambda_A \right] du$$

$$= \bar{n}_A \bar{n}_B \left[ \frac{1}{\bar{n}_A} \ln \lambda_A - \frac{1}{\bar{n}_B} \ln \lambda_B \right] du$$

so there is an extremum in  $\Delta \Omega_{sys}$

when

$$\frac{d}{du} \ln \lambda_A = \frac{d}{du} \ln \lambda_B$$

Two systems in thermal contact will exchange energy (heat) until their temperatures are equal, so we might suspect  $\frac{d}{du} \ln \lambda$  to be ~~at most~~

~~in 1-1-1 correspondence~~

~~temperature proportions at least in~~  
~~1-1-1 correspondence w/ the system's~~  
~~temperatures.~~

problem 3)

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a)

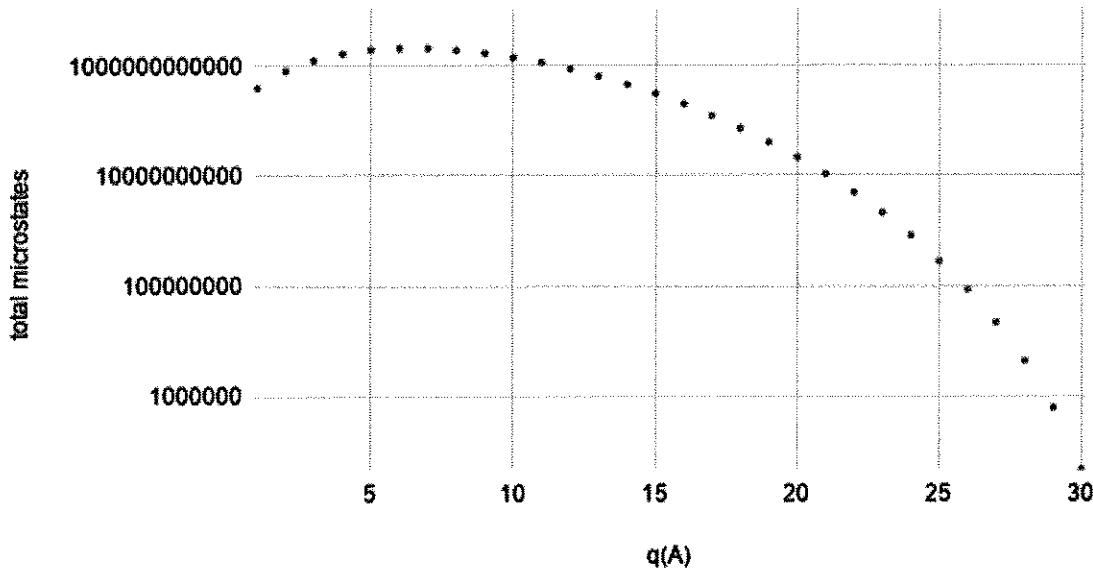
From the attached plot the total number of microstates peaks at  $q(A) = 6$

b)

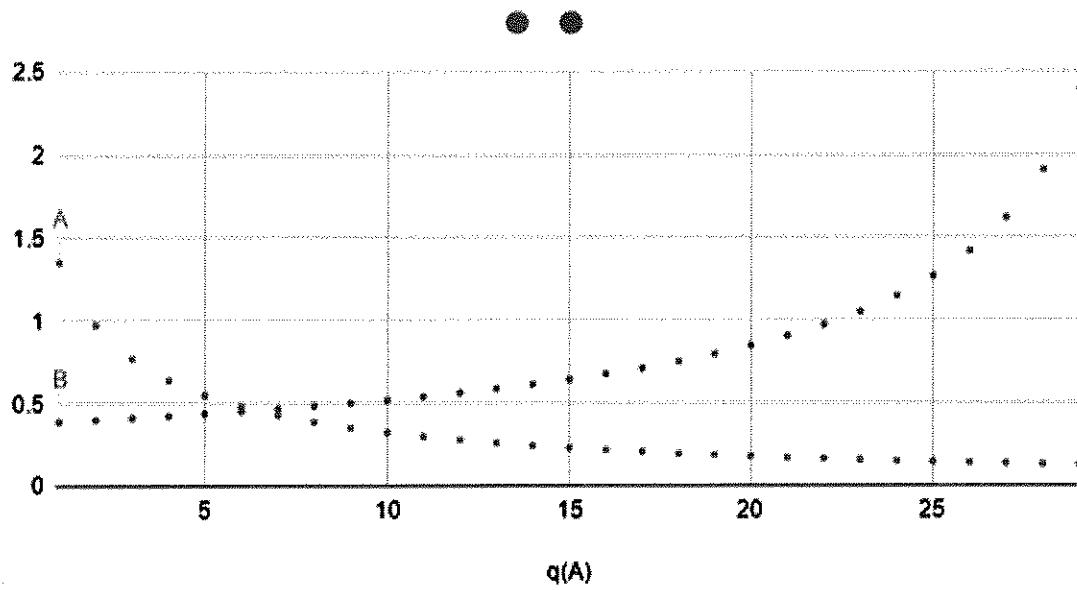
The expression for both systems is a finite analog of a derivative of the log of their number of microstates, so we might expect them to be equal when the system is in equilibrium, i.e. at  $q_{A^{MAX}}$ . This does appear to be the case.

If we interpret the derivative of the log of the microstates as something analogous to the temperature of the system, then we find that we would expect negligible change in this quantity at  $q_{A^{MAX}}$  for system A (and thus system B) upon doubling the size of system A. Though doubling the size of A at fixed  $q$  will take roughly twice as much heat away from system B, the "temperature" of system B or the slope of its log of number of microstates at equilibrium will only have changed very little because it is so much larger than A. In other words, it has a very large heat capacity.

total microstates vs.  $q(A)$



slope of log of microstates vs.  $q(A)$



4) From the symmetry of the RHS of eq : (2.17) we can  
 immediately convert the high temp  
 expression Eq (2.21) to low temp  
 by assuming  $g$  and  $\alpha$ , giving:

$$R(N, g) \approx \left(\frac{e^N}{g}\right)^T \quad g \ll N$$

For two subsystems each of size  $N$   
 splitting  $g$  energy quanta into

$$g_{1/2} + \Delta g_{1/2} \quad \text{and} \quad g'_{1/2} - \Delta g'_{1/2}$$

we have :

$$R_{\text{total}} (\Delta g) = \left( \frac{e^N}{g_{1/2} + \Delta g_{1/2}} \right) \left( \frac{e^N}{g'_{1/2} - \Delta g'_{1/2}} \right)$$

$$= \left[ \frac{2e^N}{g} (1+\varepsilon) \right]^{-1} \left[ \frac{2e^N}{g'} (1-\varepsilon) \right]^{1-\varepsilon}$$

$$\varepsilon = \frac{\Delta g}{g}$$

$$= \int \left[ \frac{2Ne}{g} \right]^{1+\varepsilon+1-\varepsilon} \left( \frac{(1+\varepsilon)^{-1}}{(1-\varepsilon^2)^{-1}} \right) \left( \frac{(1-\varepsilon)^{-1}}{\frac{(1-\varepsilon)}{(1+\varepsilon)}} \right)^{\varepsilon}$$

$$= \left[ \frac{2Ne}{g} \right]^2 \left[ \left( \frac{(1-\varepsilon)^{-1}}{\frac{(1-\varepsilon)}{(1+\varepsilon)}} \right)^{\varepsilon} \right]$$

$$\stackrel{\varepsilon(1+\varepsilon)^2}{\approx} e^{\varepsilon^2 \ln \left( \frac{(1-\varepsilon)^{-1}}{(1+\varepsilon)} \right)^{\varepsilon}}$$

$$= e^{\varepsilon \ln \left( \frac{(1-\varepsilon)^{-1}}{(1+\varepsilon)} \right)}$$

$$\approx e^{\varepsilon \ln(1-2\varepsilon)}$$

$$= e^{\varepsilon(-2\varepsilon)}$$

$$= e^{-2\varepsilon^2}$$

$$= \left( \frac{2Ne}{g} \right) / e = \left( \frac{2Ne}{g} \right) e^{-\left( \frac{\Delta g}{\sqrt{g}} \right)^2 / 2}$$

So the microstates are Gaussian distributed  
in  $\Delta g$  w/ mean of zero and standard deviation of  $\sqrt{g}$