

# Thermodynamics HW #4

Due September 23

1. Schroeder 2.8
2. Let  $A$  and  $B$  be two systems (not necessarily Einstein solids) and let  $\Omega_A(U_A)$  and  $\Omega_B(U_B)$  be the respective number of microstates available to each system (i.e. system  $A$  has  $\Omega_A(U_A)$  available microstates when it has an energy  $U_A$  and likewise for  $B$ ).

Show that the  $A$  and  $B$  as a combined system are at equilibrium when  $\frac{d}{dU_A} \ln \Omega_A = \frac{d}{dU_B} \ln \Omega_B$ .

In light of this observation, what property might be we tempted to associate with the functions  $\frac{d}{dU_A} \ln \Omega_A$  and  $\frac{d}{dU_B} \ln \Omega_B$ ?

3. Investigate the equilibrium condition established in the previous problem using a system consisting of two Einstein solids  $A$  and  $B$ , where  $N_A = 5$  and  $N_B = 15$ , and  $q = 30$ .

(a) Plot  $\Omega_A(q_A)\Omega_B(q - q_A)$  vs.  $q_A$  for  $q_A = 0, 1, 2, \dots, q$  and identify the value  $q_A^{\text{MAX}}$  for which the plot is maximum. For ease of viewing you will want to plot the y-axis on a log scale.

(b) Now construct a different plot with two curves:

- $\frac{\ln \Omega_A(q_A+1) - \ln \Omega_A(q_A-1)}{2}$ , and
- $\frac{\ln \Omega_B(q - q_A + 1) - \ln \Omega_B(q - q_A - 1)}{2}$

for  $q_A = 1, 2, \dots, q - 1$ .

Where do you expect the curves to intersect? Why? Was your expectation correct?

Now suppose  $N_B$  and  $q$  grow very large while  $N_A$  stays small. (If it helps you may also assume  $q \gg N_B$ .) In this case, how do you expect  $\frac{d}{dU_A} \ln \Omega_A(q_A^{\text{MAX}})$  to change if you double the size of system  $A$ ? Why?

4. Repeat the derivation of Schroeder equation (2.28), this time for the low temperature limit ( $q \ll N$ ). *Hint: note the symmetry of the right-hand-side of equation (2.17). You should be able to immediately write down the low temperature analog of equation (2.21) and go from there.*